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ENZYMES IMMOBILIZED BY MICROENCAPSULATION WITHIN SPHERICAL ULTRATHIN POLYMERIC MEMBRANES

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INTRODUCTION

Biological cells in nature have diameters of about 7 to 50 μ m, with a cell membrane of about 0.01 μ m in thickness. This means that the total membrane area available for diffusion can be extremely large in comparison to the volume and that the cell membrane can also be very thin without the loss of mechanical strength. In addition, rapid mixing is possible within the individual microscopic compartments. Biological cells have very complex and selective enzyme systems to act on substrates crossing the cell membrane. Artificial cells in the form of spherical ultrathin polymeric membrane envelopes containing enzymes have been prepared in this laboratory with some of these properties [1-3]. Detailed papers and upadated information published since then on this subject have been described in a recent monograph [4].

PREPARATION

The cell model system first prepared in this laboratory in 1956 consisted of red blood cell hemolysate enclosed within spherical ultrathin collodion membranes of cellular dimensions. Each "artificial cell" is of cellular dimension containing all the protein and enzyme systems of red blood cells, except that the red blood cell membrane has been replaced with an ultrathin collodion membrane. In

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this initial preparative procedure there are three main steps: (I) The aqueous protein and enzyme solution is emulsified in a waterimmiscible solution to form aqueous microdroplets. (II) With the addition of a polymer solution to the stirred emulsion, a membrane is formed on the surface of each microdroplet and then allowed to set in an appropriate solution. (III) Finally, the microencapsulated red blood cell hemolysate formed is dispersed in an aqueous phase. With modifications, this general procedure has been varied for the preparation of artificial cells with a great variety of membrane material, content, size, and configuration. The preparative procedures have been updated and described in detail [4].

Briefly, the first approaches [1,2,4] involved organic phase separation. Polymers dissolved in a water-immiscible fluid are deposited around each aqueous microdroplet dispersed in the waterimmiscible fluid. This is one of the most convenient ways of preparing microencapsulated enzymes. The details have been updated and simplified [4]. The best approach is to use the 10 g % hemoglobin solution for microencapsulation, and any enzymes to be enclosed could be added to the hemoglobin solution, since without the 10 g % hemoglobin solution, which is a basic component of the cell model, a great deal of variation and modification of the preparative procedure would be needed. Enzymes, proteins, or other material for microencapsulation are added to the 10 g % hemoglobin solution before carrying out the microencapsulation procedure.

Another approach [2-4] is to use interfacial polymerization. A diamine dissolved in the aqueous phase of the 10 g % hemoglobin reacts at the interphase of each aqueous microdroplet with sebacoyl chloride in the organic phase to form a membrane. Nylon membrane is used most commonly for this, but other membranes could also be formed, for instance, polyamides, polyureas, polyurethanes, polysulfonamides, and polyphenylesters [4]. Again, in this typical example, it is most convenient to use a 10 g % hemoglobin solution and to add to this the enzymes or other proteins that one wants to immobilize within the microcapsules. This chemical procedure, unlike the first procedure of organic phase separation, may inactivate some enzyme systems.

A third approach [4,5] would be by secondary emulsion. Here a fine emulsion of enzyme solution is dispersed in an organic phase containing a polymer. The organic phase containing the fine aqueous microdroplets is then dispersed in the form of large droplets in an aqueous phase. The organic phase containing the polymer solutions is then solidified to form solid microspheres, each containing a number of aqueous microdroplets. Unlike the two examples described earlier, this procedure results in solid polymer microspheres, each of which contains several aqueous microdroplets. A typical example is the use of silicone rubber to microencapsulate red blood cell hemolysate [5]. Other polymers include the use of cellulose membrane and other membranes [4,5]. Another procedure [7] is the emulsification of enzyme solution in a polymer organic solution, but instead of forming microspheres the polymer solution containing the enzyme microdroplets is extruded in the form of fibers. Further extensions include the microencapsulation of enzymes within liposomes [8]. However, in this case the enzyme is entrapped between onion-skin-like multiple laminar lipid. Hemolysate has also been microencapsulated within spherical ultrathin lipid membrane [9]. Enzymes have also been microencapsulated within liquid membranes [10].

Since the "artificial cell" is a general concept [4], the artificial cells prepared by the procedures that have been described are only some examples for demonstrating this concept. It is expected that modifications or extensions of these procedures can be made available to further demonstrate or improve the preparation of different types of artificial cells.

PROPERTIES

The ultrathin membranes (0.02 μm) and the large surface to volume relationship of artificial cells (2 m² area in 10 ml of 20 μm diameter artificial cells) results in extremely high transport rates across the artificial cell membranes. Thus 10 ml of these 10 μm

diameter spherical ultrathin polymer membranes has a theoretical transport capacity 200 times higher than that of a whole standard 1 m² dialysis machine [5,11]. The enclosing polymeric membrane of artificial cells is prepared in such a way that while impereable to macromolecules or suspensions, it is extremely permeable to most of the solutes normally present in the biological fluid. Enveloped by the spherical ultrathin polymer membrane is either an aqueous solution or a suspension of biologically active material like enzyme systems. While remaining at all times enveloped by the enclosing polymeric membrane and prevented from coming into direct content with the external environment, the enclosed enzyme systems act on external permeant substrates diffusing into the artificial cells. This means that enzymes cannot leak out of the spherical ultrathin polymeric membrane envelopes or come in contact with external macromolecules like antibodies. However, permeant substrates can equilibrate rapidly across the enclosing membranes to be acted on by the enclosed enzymes, and the products can diffuse out of the microcapsules. The permeability characteristics of the polymeric membranes can be varied at will as to porosity, charge, composition, and lipid contents. In addition, the polymeric membranes can also be made biologically compatible [4].

MICROENCAPSULATED ENZYMES

Table 1 lists enzymes and proteins which have been immobilized by microencapsulation. Additional pertinent references are also to be noted [16,18,32,36-38,41,49-51,53-55]. Thus hemoglobin, albumin, uricase, urease, catalase, asparaginase, carbonic anhydrase, trypsin, lactase, and many other enzymes have been immobilized by microencapsulation. Even complex enzyme systems like the contents of red blood cells and muscle cells have been microencapsulated (Figs. 1-3). The microencapsulated enzymes do not leak out but act efficiently on permeant substrates. Products formed can leave the microcapsules. Most enzyme activities are retained after microencapsulation. The kinetics of reaction are governed by a large number of factors: concentration of enzymes Enzymes and Proteins Immobilized within Semipermeable Microcapsules

	References
Enzymes and hemoglobin of red blood cell hemolysate	1-3, 6, 9, 12, 13, 28
Urease	1, 3, 5, 6, 10, 14, 15, 17, 28-31, 34, 35, 44
Carbonic anhydrase	2, 19
Uricase	2, 3
Trypsin	2
Catalase	6, 20, 42, 45, 46
Asparaginase	4, 11, 21, 22, 30, 33-35 39, 43, 47, 48, 52
Albumin	2, 3, 23
Lipase	6
Zymase complex (yeast)	24
Muscle extract	24
Phenolase	25
Nitrate reductase	26
β -Fructofuranosidase	8
β-Galactosidase	27, 39, 40, 52

in the microcapsules, permeability of membrane to the substrates and products, charge of membrane, composition of membranes, and other factors.

Enzymes can be stabilized by microencapsulation with a high concentration of protein in the form of the 10 g % hemoglobin solution [3,30,48]. Proteins cannot leak out of semipermeable microcapsules even when the microcapsules themselves are dispersed in a very dilute suspension. Thus, unlike native enzymes in a dilute solution, the microencapsulated enzyme is at all times retained in an intracellular environment with a high concentration of protein. In this way the stability of the enzyme is much greater than the enzyme in the dilute solution. For example, the half-life of



Fig. 1. Enzyme immobilized by microencapsulation within spherical ultrathin polymeric membranes. Mean diamter: 100 µm. spherical cellulose nitrate membrane of 0.02 µm thickness enveloping a complex enzyme system.

catalase in free solution is 15 days when stored at 4° C. When microencapsulated with a high concentration of protein and stored at 4° C, it retained more than 90% of its initial activity for more than 100 days. The half-life of asparaginase in free solution was 20 days when stored at 4° C, but when microencapsulated with a high concentration of protein the enzyme retained 100% of its initial activity for more than 100 days [4,39,48]. Urease and trypsin also become more stable after microencapsulation with a high concentration of protein. Enzymes can also be stabilized by directly microencapsulating insolubilized enzymes [4,34]. An easier way to insolubilize microencapsulated enzymes is to microencapsulate the enzyme at a very high concentration. Following this, it is treated



Fig. 2. Schematic representation of the preparation of spherical ultrathin polymeric membranes containing (a). (b) and (c) are modifications and extensions to form solid spheres (b) or fibers (c).

with glutaraldehyde, thereby cross-linking the enzymes within the microcapsule [48]. In this way, microencapsulated catalase and asparaginase retained more than 90% of their original activity after storage for more than 100 days at 4°C [48]. At body temperature of 37°C, the microencapsulated enzymes are also more stable.



MICROENCAPSULATED ENZYME

Fig. 3. Schematic representation of the action of microencapsulated enzymes. Enzyme does not leak out, but acts on permeant substrates. Antibody cannot enter to adversely affect the enzyme.

Microencapsulated enzymes have been injected intraperitoneally or used in extracorporeal shunts [4]. These microencapsulated enzymes did not leak out of the microcapsules but continued to act on substrates diffusing into the microcapsules. Urease acted on body urea [4]; microencapsulated catalase acted efficiently in replacing the deficient enzyme in an inborn error of metabolism; acatalasemia [4] and microencapsulated asparaginase acted efficiently in suppressing the growth of an asparagine-dependent lymphosarcoma [4]. Microencapsulated enzymes remain inside the artificial cells after injection. They are not inactivated by cells or enzymes in the body and are not removed rapidly. As a result, injected microencpasulated enzymes, such as catalase and asparaginase, acted significantly longer than when the same enzymes were injected in

TABLE 2

Stages of Enzyme Research

Ι.	Research on enzymes in dilute solution.
II.	Research on enzymes immobilized by covalent linkage, adsorption, or gel-entrapment as models of membrane- bound enzymes.
111.	Research on enzymes immobilized by microencapsulation within spherical ultrathin polymer membranes as models of intracellular enzymes (artificial cells).

the free form [4]. Furthermore, microencapsulated enzymes, unlike free enzymes, are prevented from taking part in immunological reactions [4,42].

IMMOBILIZED ENZYMES

RELATIVE AMOUNT OF MATRIX SUPPORT (===) AND ENZYME (E)



Fig. 4. Stages of enzyme research, illustrating the relative amounts of matrix support () and enzyme (E). Stage II: matrix entrapped, adsorbed, and covalently bound. Stage III: microencapsulated.

GENERAL

Enzyme research has proceeded from dilute enzyme solutions to the second level of gel-entrapped, adsorbed, and covalently bound enzymes as a model for membrane-bound enzymes, and finally, with the application of advances in polymer chemistry, we have demonstrated the feasibility of proceeding to the level of artificial cells containing enzymes as models of biological cells (Table 2, Fig. 4). With the experimental demonstration of the possible application of this spherical ultrathin polymeric system for basic research, applied research, and clinical use, further efforts are indicated, especially to apply the enormous amount of knowledge in polymer chemistry to this system.

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